Nickel-catalyzed [4 + 2] Cycloaddition of Alkynes to Carbonylsalicylamides via Elimination of Isocyanates

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An intermolecular nickel-catalyzed [4+2] cycloaddition has been developed where carbonylsalicylamides react with alkynes to afford substituted chromones. The reaction involves an elimination of isocyanates and an addition of alkynes, that is, a formal substitution reaction of an isocyanate by an alkyne.

Addition of carbon and heteroatom across unsaturated carbon-carbon bonds to form carbon-carbon and carbon-heteroatom bonds simultaneously would be a potentially useful and straightforward method for introducing a heteroatom function into organic molecules.^{1,2} Recently, we demonstrated that an oxidative addition of an isatoic anhydride to a Ni(0) complex and subsequent decarboxylation afford an aza-nickelacycle A; the process leads to the carboamination of alkynes to provide quinolon in a single step (Scheme 1a).³ We postulated that an oxa-nickelacycle B might be generated alternatively via an oxidative addition to Ni(0) of carbonylsalicylamide 1, which is a structural isomer of isatoic anhydrides, and elimination of isocyanate (Scheme 1b). This would allow forming carboncarbon and carbon-oxygen bonds simultaneously and would open the way for divergent synthesis of chromone $3^{4,5}$ Thus, we attempted the [4+2] cycloaddition of alkyne 2 to carbonylsalicylamide 1 using Ni(0) catalyst to form chromones 3.

Our investigation began with an attempted addition of 4-octyne (2a) to N-phenylcarbonylsalicylamide (1a) with 10 mol % of Ni(cod)₂ and 30 mol % of PCy₃ in *i*-BuCN for 24 h. However, this resulted in low conversion of 1a and led to chromone 3aa in 46% yield (Table 1, Entry 1).⁶ A detailed examination of the reaction revealed that nickel-catalyzed [2 + 2 + 2] cycloaddition of 2a and isocyanate, which is generated during the addition reaction, was a side reaction that prohibited the catalytic reaction (Scheme 2).⁷ Thus, we investigated the inactivation of isocyanate, and found that, on addition of 10 mol % of 1,3-bis(2,6-diisopropylphenyl)-4,5-dihydroimi-

Scheme 1.

dazol-2-ylidene (SIPr), the desired reaction proceeded to furnish **3aa** in 59% yield (Entry 2).⁸ The effect of SIPr is likely to result from promoting the cyclotrimerization of isocyanate to afford an isocyanurate,⁹ which is inert to oxidative addition to nickel-catalyst. In the case of using SIPr as a single ligand for nickel catalyst, **3aa** was not obtained (Entry 3). This can be rationalized that PCy₃ is an essential ligand for the cycloadditon of **2** to

Table 1. Nickel-catalyzed $\left[4+2\right]$ cycloaddition of carbonylsalicyl-amides to alkynes^a

	O Ph				Ni(cod) ₂ (10 mol%) PCy ₃ (30 mol%) SIPr (10 mol%)			O R ¹	
R			+	R ¹ 2 equi	—R ² — 011 iv <i>i</i> -B	BuCN, 160 °C, 24 h R			
Entry	1	1 R	2	2 R ¹	R ²	Conversion/% ^b	3	3 Yield/% ^c	
1	19	н	2.9	Pr	Pr	73	399	46 ^d	
2	1a	Н	2a	Pr	Pr	97	3aa	59	
3	1a	Н	2a	Pr	Pr	19	3 aa	<1 ^e	
4	1a	Н	2b	Bu	Et	75	3ab	55 (1/1) ^f	
5	1a	Н	2c	<i>i</i> -Pr	Me	68	3ac	$50(1/1)^{f}$	
6	1a	Η	2d	t-Bu	Me	54	3ad	28	
7	1a	Η	2e	Me ₃ Si	C_6H_{13}	61	3ae	55	
8	1b	CF_3	2a	Pr	Pr	98	3ba	80	
9	1b	CF_3	2b	Bu	Et	93	3bb	$52 (1/1)^{f}$	
10	1b	CF_3	2c	<i>i</i> -Pr	Me	82	3bc	$44 (1/1)^{f}$	
11	1b	CF_3	2d	t-Bu	Me	58	3bd	42	
12	1b	CF_3	2e	Me ₃ Si	C_6H_{13}	94	3be	77	
13	1c	MeO	2a	Pr	Pr	33	3ca	21	
14	1c	MeO	2b	Bu	Et	46	3cb	$12 (1/1)^{f}$	
15	1c	MeO	2c	<i>i</i> -Pr	Me	39	3cc	$11 (1/1)^{f}$	
16	1c	MeO	2d	t-Bu	Me	38	3cd	18	
17	1c	MeO	2e	Me ₃ Si	$\mathrm{C}_{6}\mathrm{H}_{13}$	81	3ce	46	

^aReactions were carried out using Ni(cod)₂ (10 mol %), PCy₃ (30 mol %), SIPr (10 mol %), **1** (0.5 mmol), and **2** (1 mmol) in 1 mL of *i*-BuCN at 160 °C for 24 h. ^bDetermined by ¹H NMR spectroscopy of crude reaction mixture. ^cIsolated yields. ^dReaction was carried out without addition of SIPr (10 mol %). ^cReactions was carried out without addition of PCy₃ (30 mol %). ^fRatio of regioisomers. Determined by ¹H NMR spectroscopy of crude reaction mixture.



Scheme 2.

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Scheme 3.

1, while SIPr promotes cyclotrimerization of eliminated isocyanate to provide isocyanurate.

With the optimized conditions in hand, we investigated the scope of the reaction using various alkynes. The reaction of 1a with unsymmetrical alkynes such as 2b and 2c gave the products consisting of regioisomers in a 1/1 ratio in moderate yields (Entries 4 and 5). The reaction of 1a with 2d, containing sterically hindered groups, gave adduct 3ad as a single product (Entry 6). Bulky trimethylsilyl-substituted alkyne 2e reacted with 1a to provide adducts with complete regiocontrol in 55% yield (Entry 7). However, aryl-substituted alkyne and terminal alkynes failed to participate in the reaction. The cycloadditions of alkyne to various carbonylsalicylamides 1 were also examined. The reaction of 2a with carbonylsalicylamides possessing an electron-withdrawing group 1b afforded the correspondingly substituted chromone 3ba in 80% yield (Entry 8). The reaction of 1b with various alkynes provided chromones in moderate yields (Entries 9-12). The reaction of 2a with 1c, which possesses an electron-donating group, afforded 3ca in 21% yield (Entry 13). The low reactivity of 1c in the cycloaddition may be ascribed to restrictions on an oxidative addition of 1c to nickel catalyst.

We propose a plausible mechanism involving oxidative addition of carbonylsalicylamide **1** to Ni(0) having electronrich phosphine ligands, giving the nickelacycle **4** (Scheme 3). Subsequent elimination of isocyanate through β -oxygen elimination provides oxanickelacycle **5**.¹⁰ Insertion of **2** to C–Ni bond leads to the seven-membered nickelacycle **7**, which undergoes reductive elimination to give **3** and regenerates the starting Ni(0) complex, while eliminated isocyanate was cyclotrimerized by SIPr catalyst during the reaction to form isocyanurate.

In summary, we have developed a new nickel-catalyzed [4+2] cycloaddition of alkynes to carbonylsalicylamides to provide chromones. It was demonstrated for the first time that isocyanate is capable of β -elimination, which allows formation of the key oxanickelacycle intermediate.¹¹ Further studies for explanation of the detailed mechanism are currently under investigation in our laboratories.

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